Global challenges of capturing carbon dioxide

Stefano Brandani*, Enzo Mangano

School of Engineering, The University of Edinburgh, Edinburgh, UK *s.brandani@ed.ac.uk

It is now widely accepted that climate change is one of the most important global challenges that we face [1]. What is also widely accepted is that no single solution is available, but a concerted effort should be made to try and limit the impact of global warming by switching to renewable energy generation and reducing emissions of carbon dioxide and other greenhouse gases. These are not new concepts, especially if one considers that more than 20 years have passed from the Kyoto protocol of 1997 [2].

While most of the focus has been on reducing emissions from electricity generation, much less emphasis has been given to the fact that these emissions typically represent only one third of the overall emissions and with further growth of renewable electricity generation this fraction is bound to decrease. The real challenge lies in decarbonizing transport, industry and distributed heating, all of which represent systems that are very different from large scale thermo-electric power plants.

Within this general context, this talk will consider the use of novel nanoporous materials as the basis for adsorption based separations [3] that will range from concentrated mixtures to direct capture of carbon dioxide from air. An overview of different classes of materials will show how these can be tailored to such a wide range of conditions. The sheer scale of the task leads to having to optimize systems and speed up processes, which in turn brings in diffusion limitations.

The final part of the talk will highlight the different types of diffusion mechanisms [4] that impact on carbon capture separations. This is a very rich field if one takes into account the range of novel materials that have been developed in the last 15 years and of particular interest are materials that undergo structural flexibility during adsorption. The discussion will focus on how to exploit diffusion limitations to enhance selectivity towards carbon dioxide and also when diffusion limitations should be minimized.

References

- [1] United Nations / Framework Convention on Climate Change (2015): *Adoption of the Paris Agreement*. 21st Conference of the Parties, Paris: United Nations.
- [2] United Nations / Framework Convention on Climate Change (1997): Kyoto Protocol to the United Nations Framework Convention on Climate Change. 3rd Conference of the Parties, Kyoto: United Nations.
- [3] J. C. Abanades, B. Arias, A. Lyngfelt, T. Mattisson, D. E. Wiley, H. Li, M. T. Ho, E. Mangano, S. Brandani: *Emerging CO₂ capture systems*. International Journal of Greenhouse Gas Control. **40**, 126–166 (2015).
- [4] J. Kärger, D. M. Ruthven, D. N. Theodorou: *Diffusion in nanoporous materials*. Wiley-VCH, Weinheim, (2012).

Stefano Brandani is the Chair of Chemical Engineering at the University of Edinburgh since 2007, having previously been Professor of Chemical Engineering at University College London. His main research interests are the fundamentals of adsorption and adsorption processes; he has over 160 publications addressing the simulation of adsorption processes and experimental measurement of adsorption and diffusion, including zero length column (ZLC) measurements for rapid ranking of novel adsorbents. He is the recipient of a Philip Leverhulme Prize and a Royal Society-Wolfson Research Merit Award. He is a Fellow of the IChemE and a senior member of the AIChE. He is Cavaliere dell'Ordine della Stella d'Italia for services to



scientific and technological research. For more information on the Carbon Capture group at the University of Edinburgh visit https://www.carboncapture.eng.ed.ac.uk/