

# Mass Diffusivities of Binary Mixtures of Normal Alkanes with Dissolved Gases

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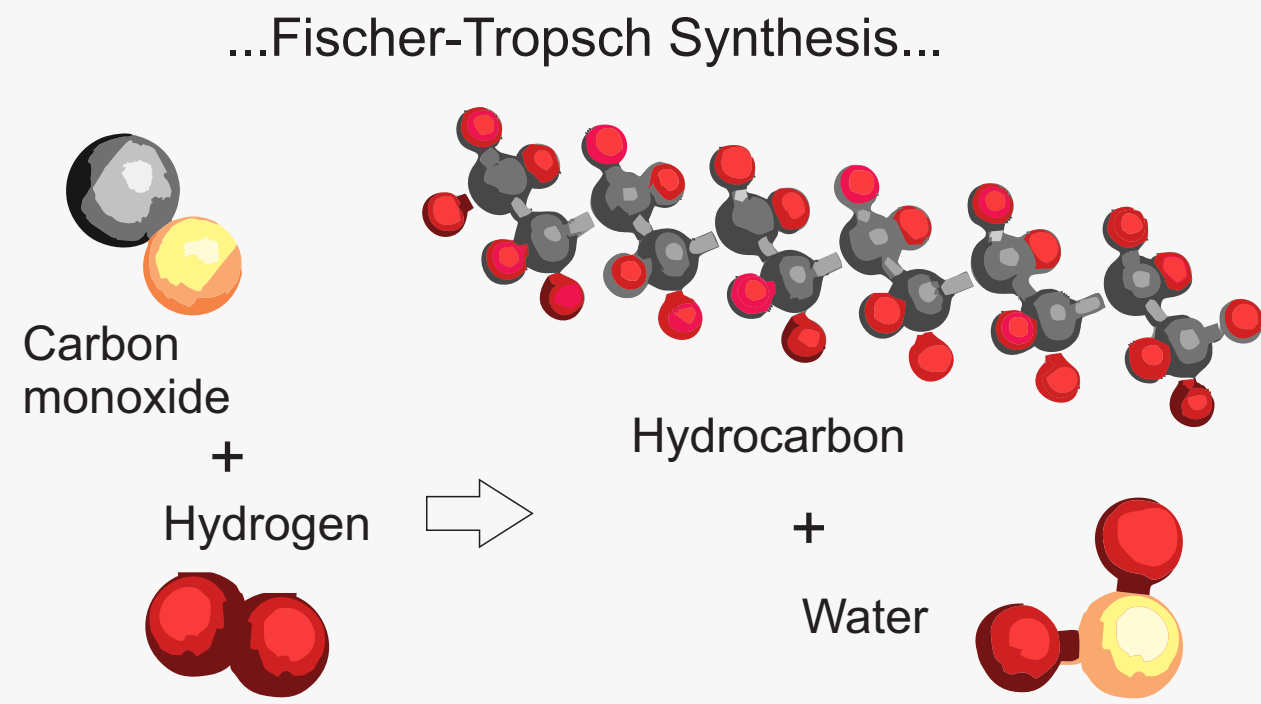
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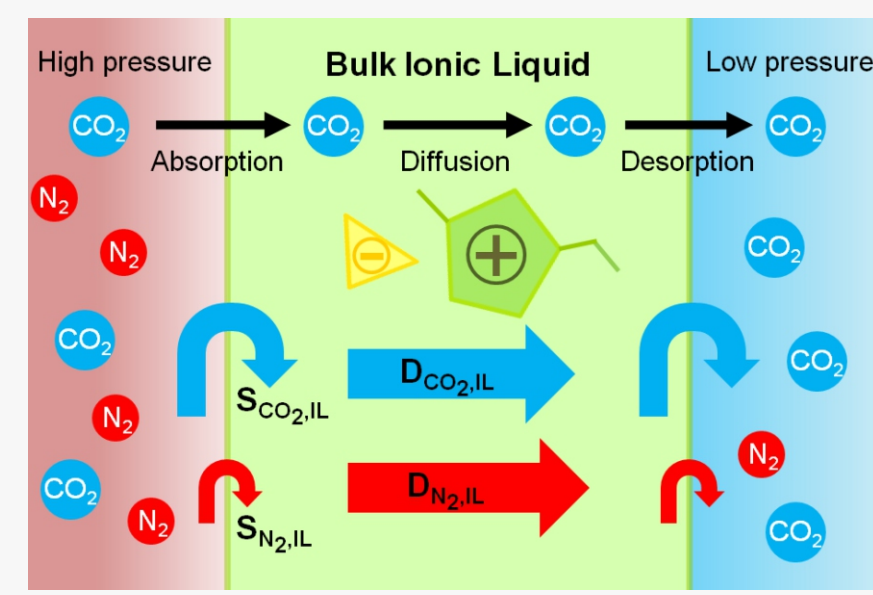
## Motivation

For which applications is the investigation of mass diffusion coefficients for liquids with dissolved gases of interest?

Example systems relevant for ...



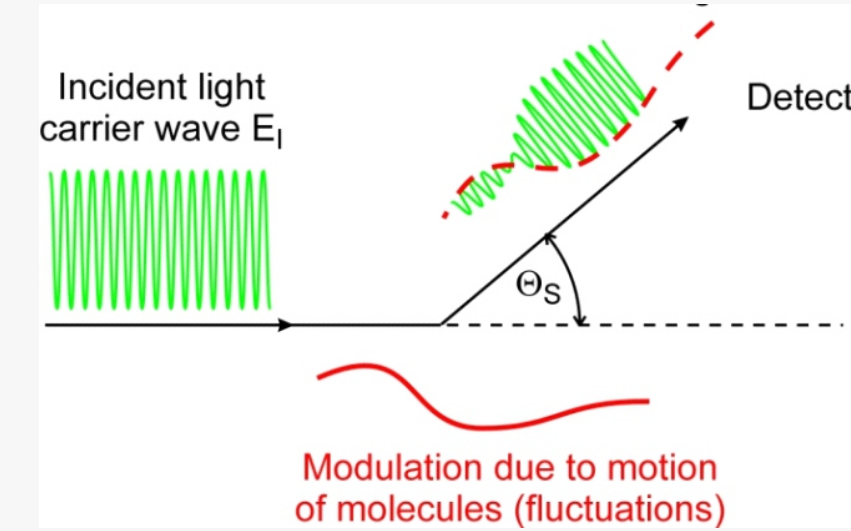
... and Supported Ionic Liquid Membranes



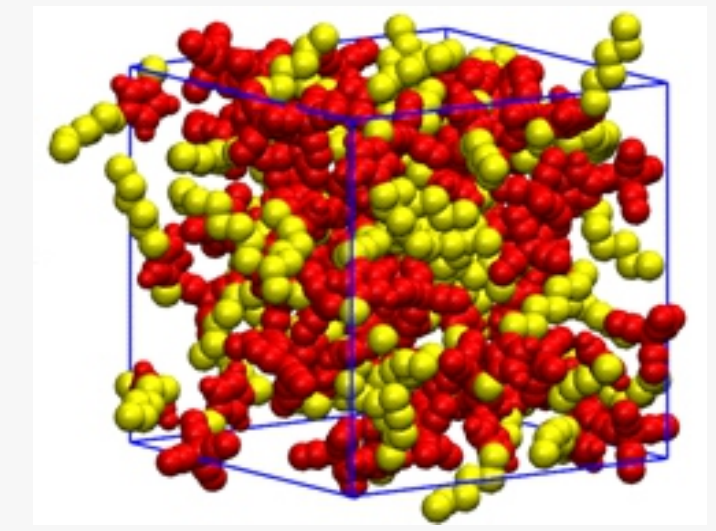
Which techniques are preferable for the investigation of mass diffusion coefficients?

Methods studying microscopic fluctuations in macroscopic thermodynamic equilibrium

Dynamic light scattering (DLS)



Molecular Dynamics (MD) simulations

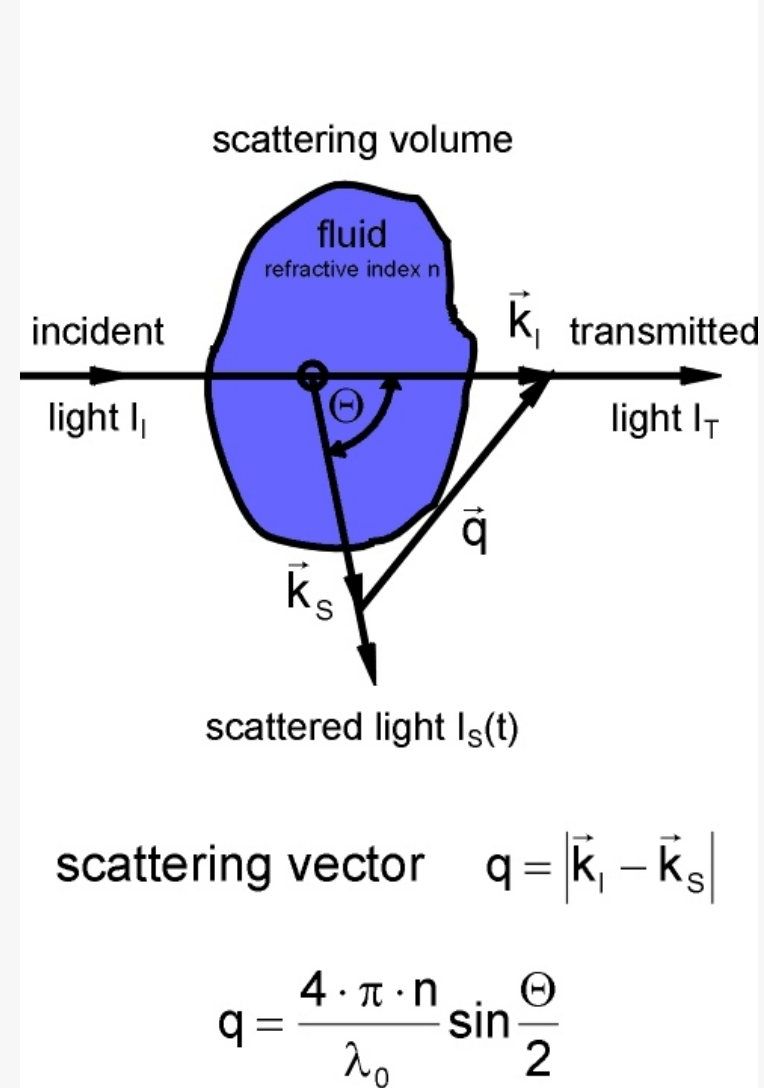


validation  
microscopic insight

## Dynamic Light Scattering (DLS)

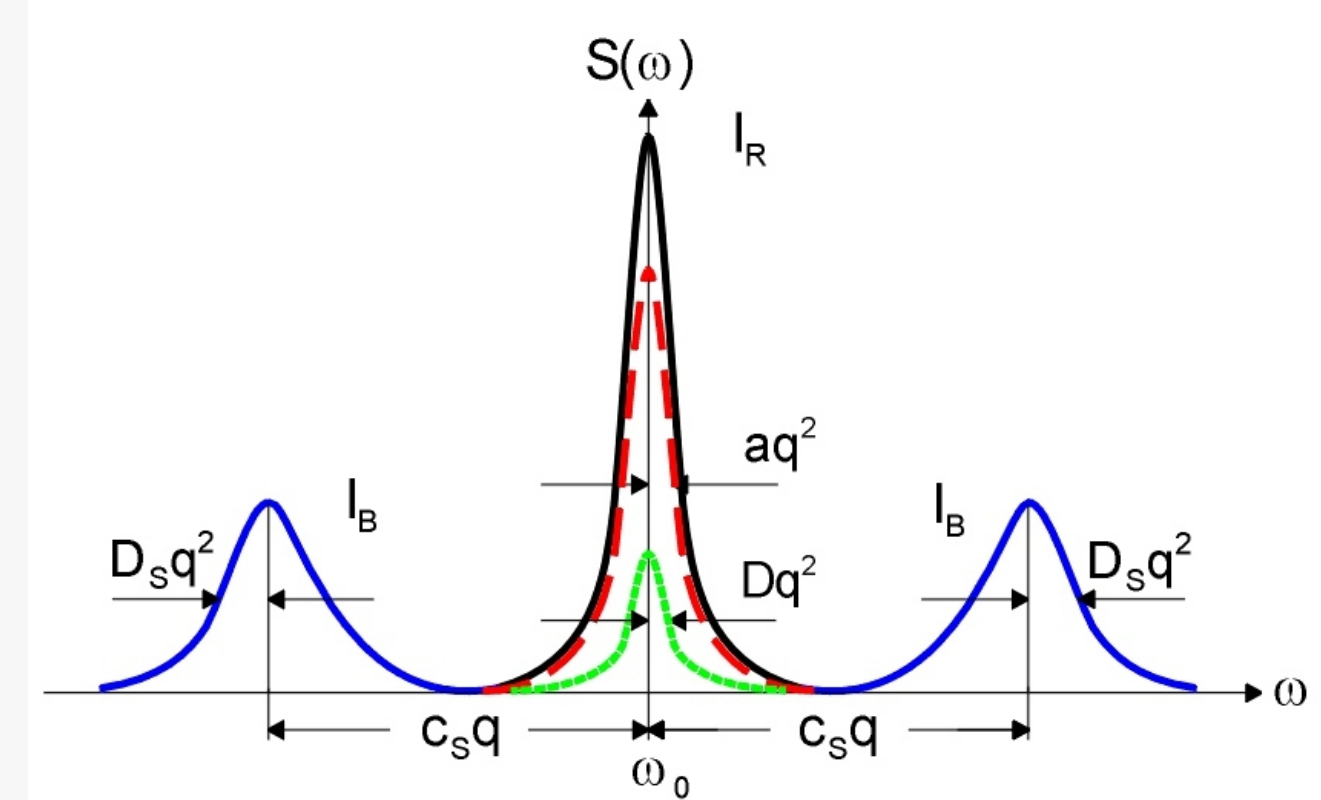
Principle

Scattering Geometry



Spectrum of Scattered Light

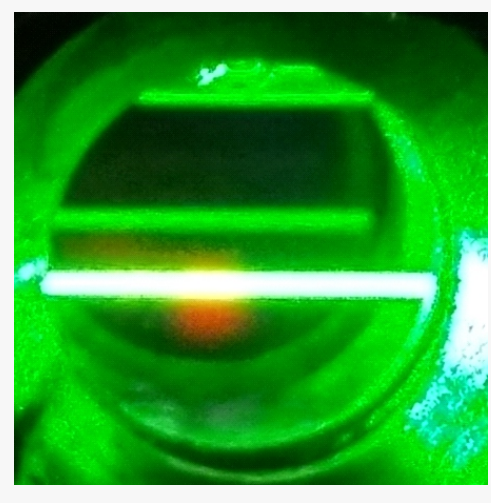
caused by local statistical fluctuations of temperature, concentration, and pressure in thermodynamic equilibrium



(a: thermal diffusivity; D: diffusion coefficient; c<sub>s</sub>: sound velocity; D<sub>s</sub>: sound attenuation; S: Landau-Placzek ratio)

Data evaluation

Visual observation during sample preparation



Pure *n*-alkane

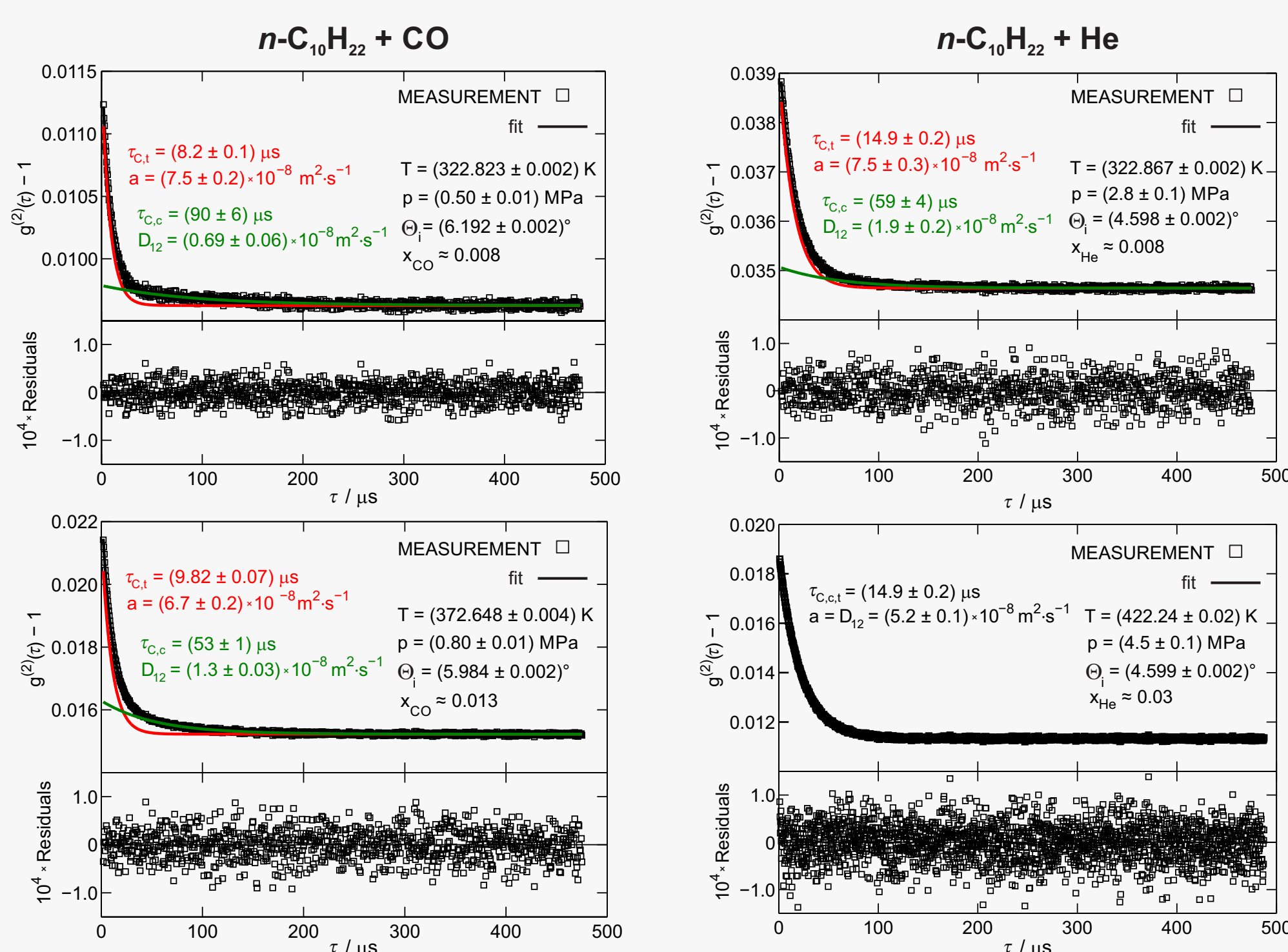


10 min after adding CO

$$Le = \frac{a}{D_{12}}$$

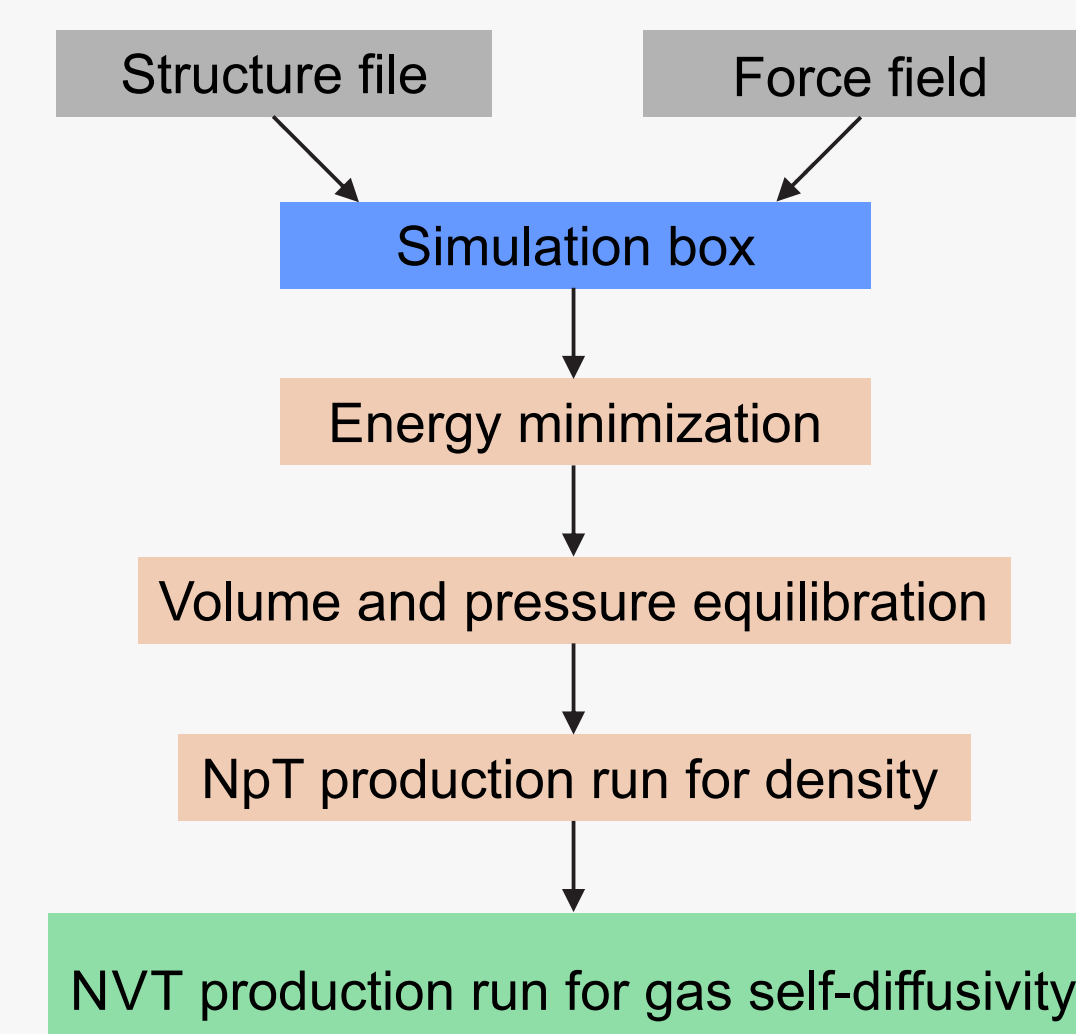
Representation of the correlation functions by a sum of two exponentials

$$g^{(2)}(\tau) = b_0 + b_1 \exp(-\tau/\tau_{c1}) + b_2 \exp(-\tau/\tau_{c2})$$



## Molecular Dynamics (MD) Simulations

Principle



Force field

$$U = \sum_i \sum_j \frac{1}{4\pi\epsilon_0} \frac{q_i q_j}{r_{ij}} \quad \text{Electrostatic potential}$$

$$+ \sum_i \sum_j \sum_k \left[ \frac{\sigma_{ij}}{r_{ij}}^{12} - \frac{\sigma_{ij}}{r_{ij}}^6 \right] \quad \text{Lennard-Jones (LJ) potential}$$

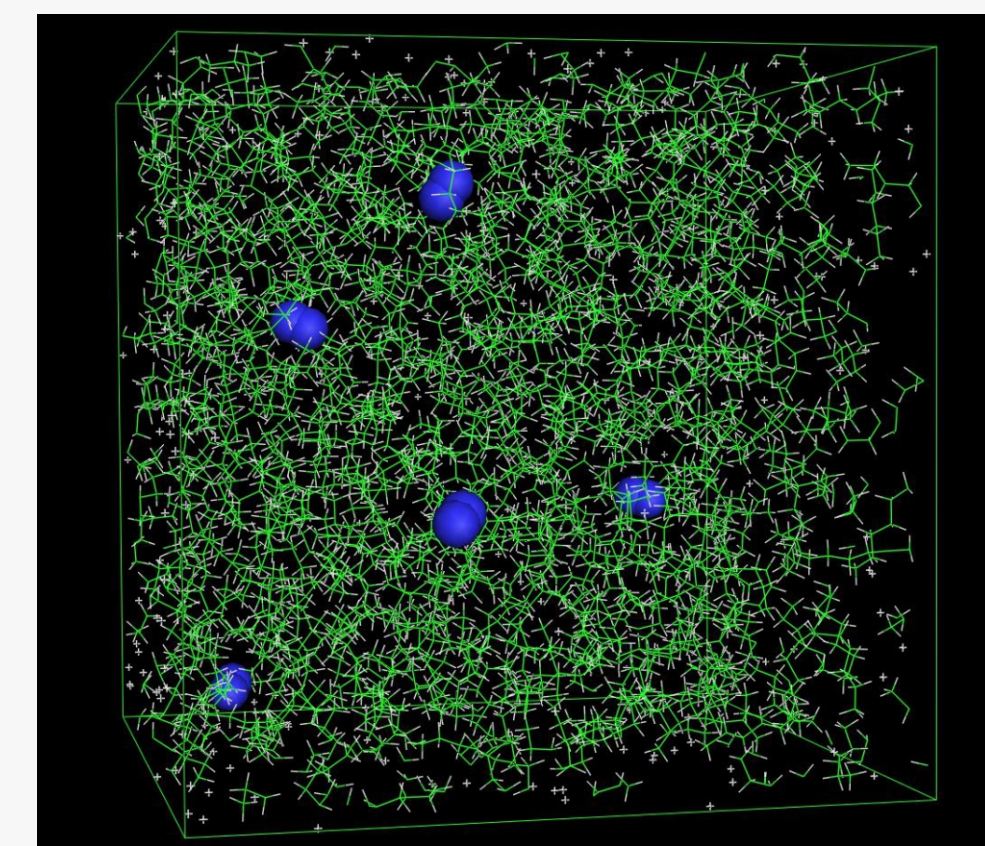
$$+ \sum_{\text{bond}} \frac{k_r}{2} (r - r_0)^2 + \sum_{\text{angle}} \frac{k_\theta}{2} (\theta - \theta_0)^2 + \sum_{\text{torsion}} k_\tau [1 + \cos(n\chi - \delta)] \quad \text{Intramolecular potential}$$

Simulation details

- Force fields for *n*-alkanes and gases from literature
- 70 ns production runs for gas self-diffusivities
- 5 independent simulation runs
- Mole fraction of dissolved gas 1%

Data evaluation

Simulation box containing 500 *n*-C<sub>10</sub>H<sub>22</sub> (1) and 5 N<sub>2</sub> (2) molecules (edge length of 4.83 nm)



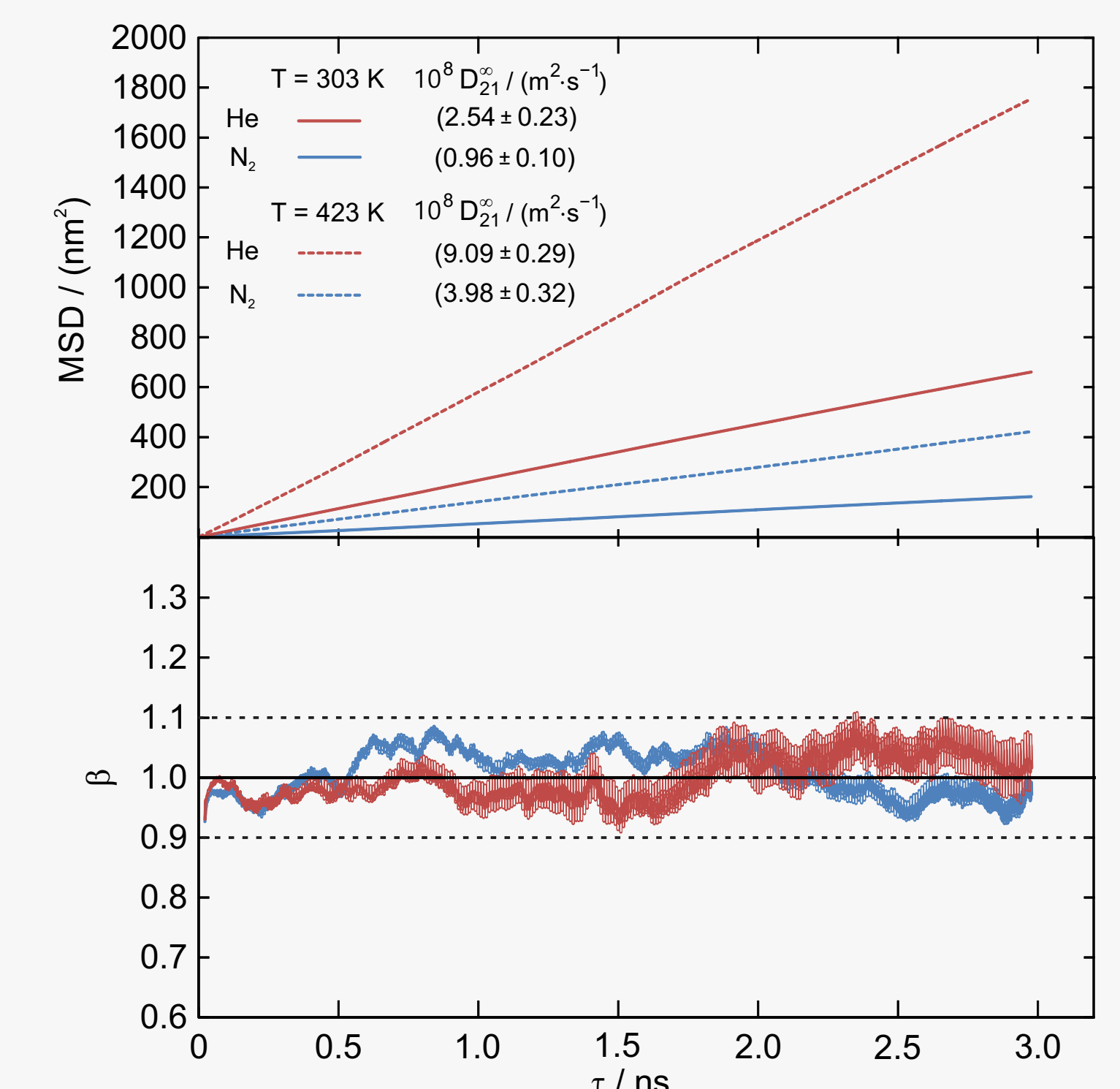
$$\text{coefficient } \beta = \frac{d(\log(\text{MSD}_2))}{d(\log(\tau))}$$

with  $\beta = 1$  for Fickian diffusive regime

Gas self-diffusivity by fit in the linear part ( $\beta = 1$ ) according to the Einstein equation

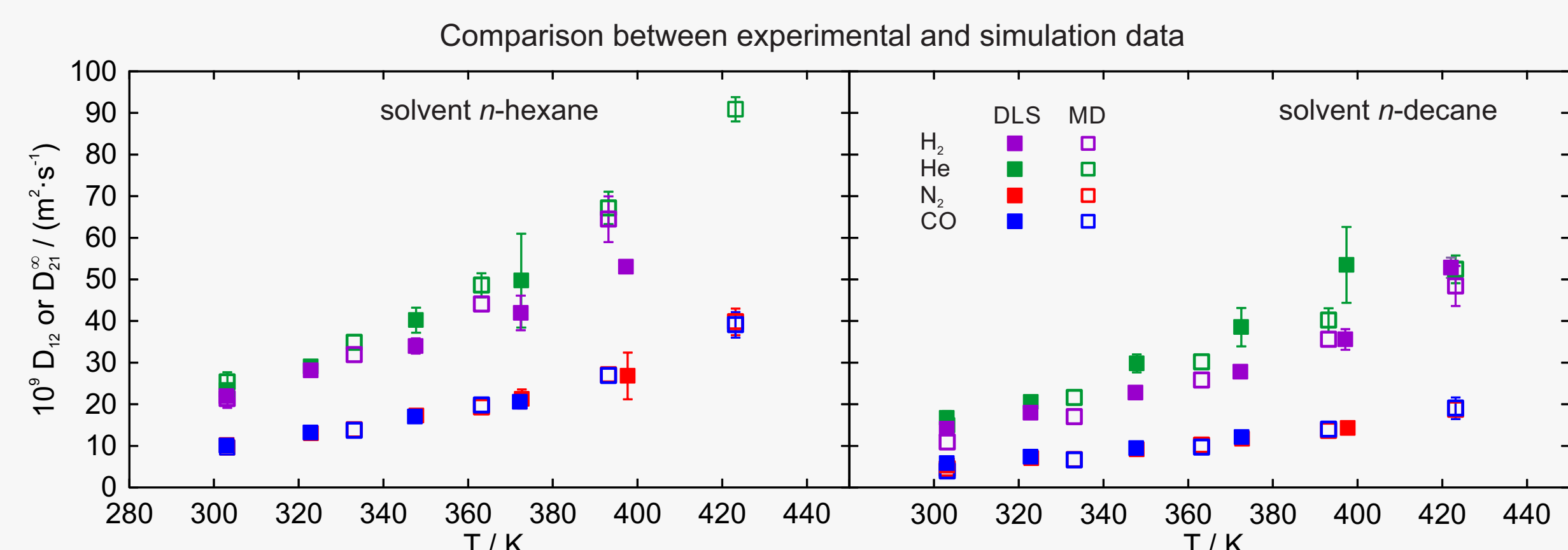
$$D_{21}^* = \frac{1}{6} \lim_{\tau \rightarrow \infty} \frac{d(\text{MSD}_2)}{d\tau}$$

Mean square displacement (MSD)  $\langle (r_2(t) - r_2(0))^2 \rangle$  of gases dissolved in *n*-hexane by 1 mole-%



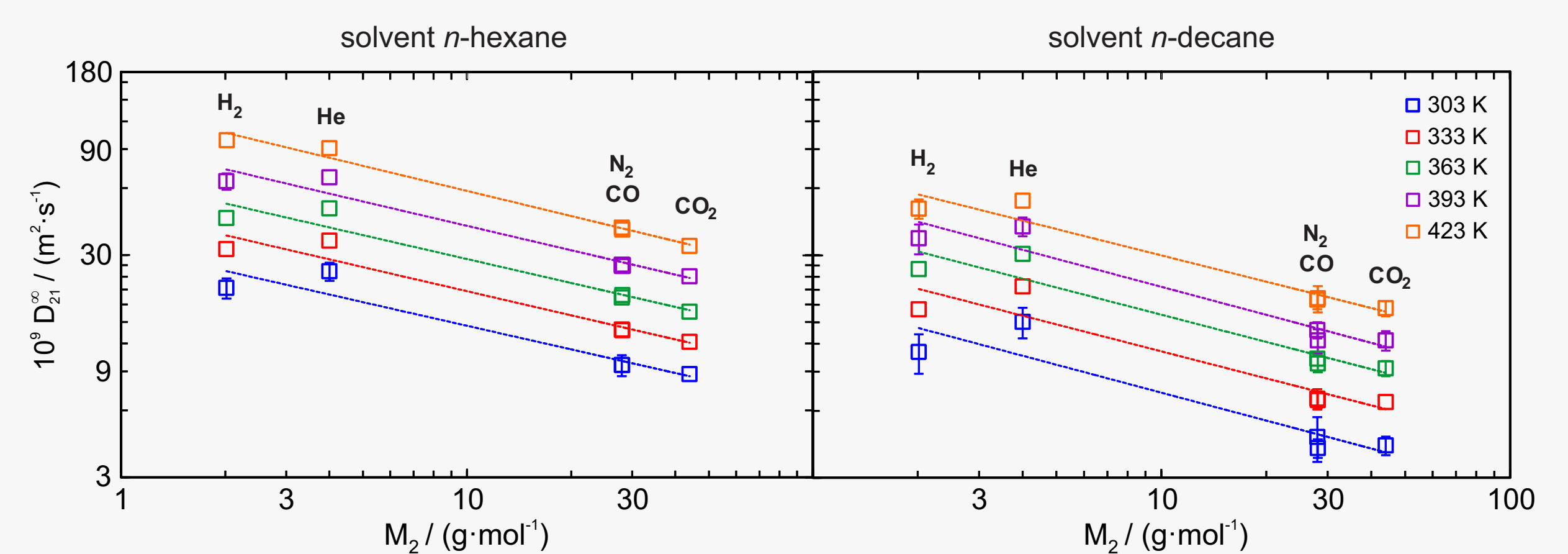
## Results and Discussion

Mass diffusion coefficients for binary mixtures of *n*-alkanes with dissolved gases in dependence on temperature close to infinite dilution



In agreement with theory, similar values for the mutual diffusivity  $D_{12}$  and the self-diffusivity of the gas  $D_{21}^*$

Self-diffusion coefficients of gases for binary mixtures of *n*-alkanes with dissolved gases in dependence on molar mass of the gaseous solute



With exception of H<sub>2</sub> and He, decreasing mass diffusion coefficients with increasing molar mass of gaseous solute

## Prospects

- Study of two further classes of alkane-based solvents: normal alcohols and ionic liquids
- Using DLS data to test the capabilities of MD simulations for the prediction of mutual diffusivities
- Development of simple prediction scheme for mutual diffusivities of liquids with dissolved gases

## Acknowledgements

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